# Characterization of Coupled Hydrologic-Biogeochemical Processes using Geophysical Data

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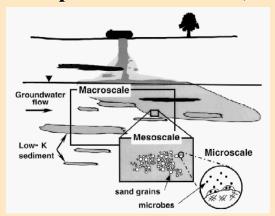
- \* Project Conception: Problem Statement & Previous Research
- EMSP Project 86922: New Start
  - Objectives
  - Current Experiment
  - Expected Project Results

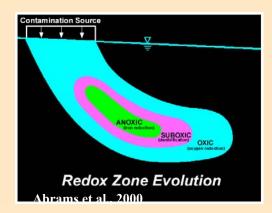




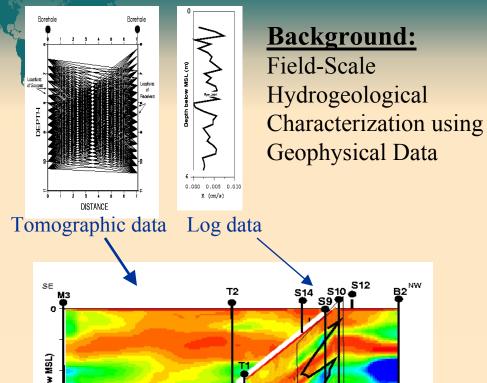
#### Statement of the Problem

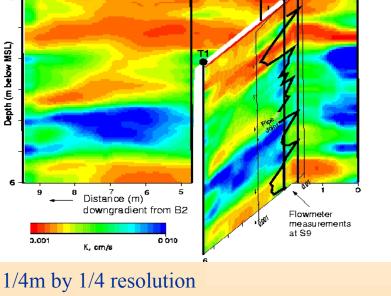
Large spatial and temporal variability of hydrogeological-biogeochemical parameters in natural systems renders characterization/monitoring of subsurface processes difficult;

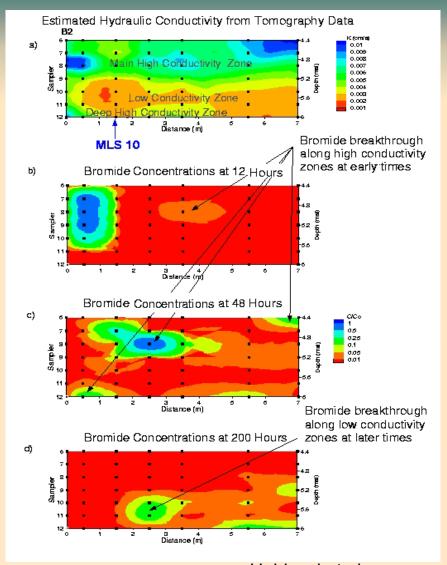




- Conventional field characterization and monitoring techniques are often inadequate:
  - Difficulty in getting information about **horizontal** or **field-scale** variability;
  - Requires drilling, which disturbs subsurface and is prohibitive in contaminated areas.
- **◆** As a consequence, characterization/monitoring is insufficient at the field scale:
  - Remediation schemes can be unnecessarily expensive, ineffective, or difficult to assess away from treatment well(s)
  - Inadequate data available to investigate **coupled processes** *AT THE FIELD SCALE*.
- **◆ Investigate potential of geophysical methods for monitoring system transformations and for understanding coupled processes.....**





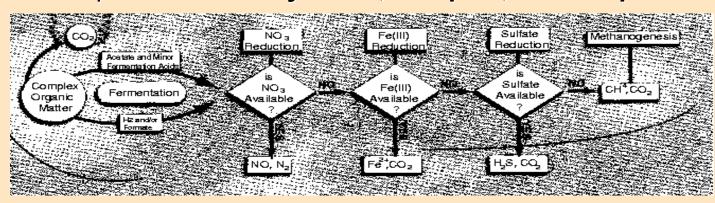


Hubbard et al., WRR 37(10), 2001

\*Tomography Estimates useful for **improving chemical transport predictions** (Scheibe et al., Ground Water, 2003) and for **understanding bacterial transport** as a function of hydrogeological heterogeneity (Mallioux et al., WRR, 2003).

# Moving Forward: Monitoring System Transformations during Remediation using Geophysical Data

- Many remediation approaches also induce system transformations (gas, biofilm and precipitation formation)
- These processes are dynamic, complex, and coupled



- Extremely difficult to understand using wellbore data
- Investigate utility of geophysical methods for providing information about system transformations over space and time
  - \* Radar, Seismic and Electrical Methods
  - Lab and Field Scale



Chapelle,

2000

## Expected Geophysical Signatures associated with Reaction Products & Hydrogeological Heterogeneity

#### Examples of Reactions associated with remediation treatments:

- Organic Carbon + Electron Acceptor + Nutrient  $\Rightarrow$  Biomass + CO<sub>2</sub> + other
- DOC + NO<sub>3</sub>  $\Rightarrow$  CH<sub>2</sub>O (biomass) + CO<sub>2</sub> + N<sub>2</sub>
- Chlorinated Solvent + Permanganate  $\Rightarrow$  Precipitate + CO<sub>2</sub> +Anions
- DOC +  $SO_4 \Rightarrow Biomass + Sulfide$

#### Examples of geophysical responses to reaction products and hydrogeology:

| Geophysical<br>Method | Geophysical<br>Attribute | Gas<br>Evolution | Precipitation<br>Formation | Biofilm<br>Formation |
|-----------------------|--------------------------|------------------|----------------------------|----------------------|
| Radar                 | Velocity                 | Increase         | Increase                   | Increase (?)         |
|                       | Amplitude                | Increase         | Variable                   | Increase?            |
| Seismic               | Velocity                 | Decrease         | Increase                   | Increase             |
|                       | Amplitude                | Decrease         | ?                          | ?                    |
| Electrical            | Conductivity             | Decrease         | Variable                   | Increase             |







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### Preliminary Lab Experiment: using Seismic Amplitudes to detect gas generation

#### 



\*Cross-column seismic measurements

Evaluate seismic signal amplitude as a function of gas production



- \*K measured using constant head tests
- \* Gas sampling of evolved N<sub>2</sub>

Electron Donor: Nitrate

Carbon Source: Acetate

#### Microbe:

Pseudomonas Stutzeri (courtesy PNNL)

Grown to  $\sim 2 \times 10^7$  cells/gram sand and suspended in a nutrient depleted growth media

Ken Williams with assist from Mary Firestone (UCB) and Fred Brockman (PNNL). In preparation for ES&T

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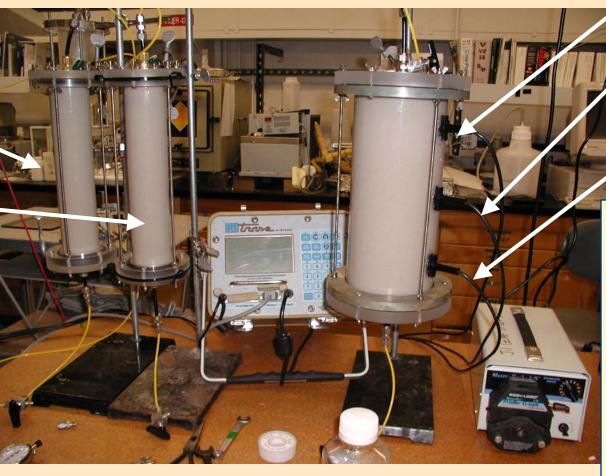


# Preliminary Radar Experiment: Monitoring of Gas Evolution during biostimulation of OY-107

Probe 1

Ksat

Seismic Columns



Probe 2

Probe 3

Electron
Acceptor:
Nitrate, Initial
Concentration

~300mg/L

Carbon Source:
Acetate

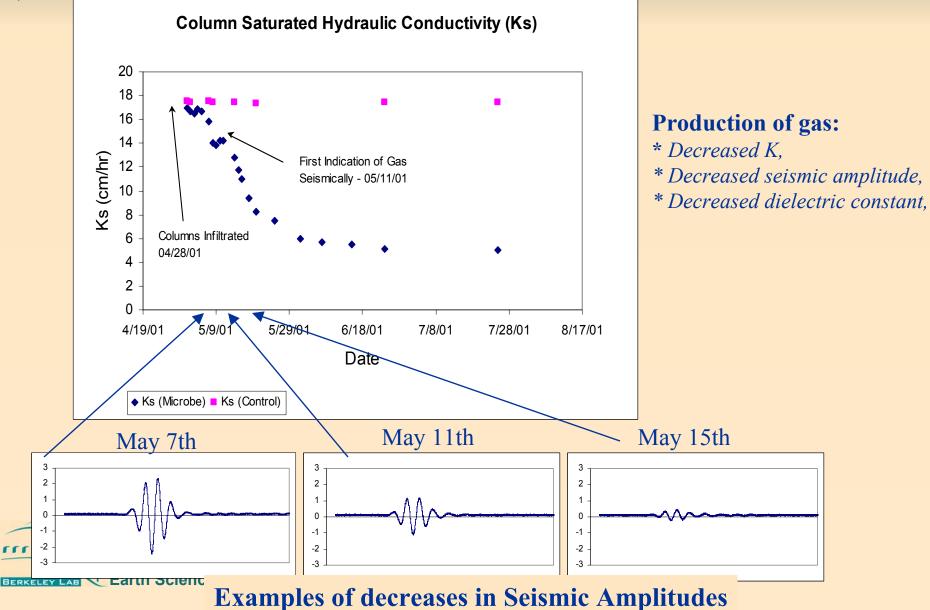
Microbe: OY107
Acidovorax

Grown to  $\sim 2 \times 10^7$  cells/gram in sand and suspended in a nutrient depleted growth media

Dielectric measurements collected every hour at each probe for 40 days



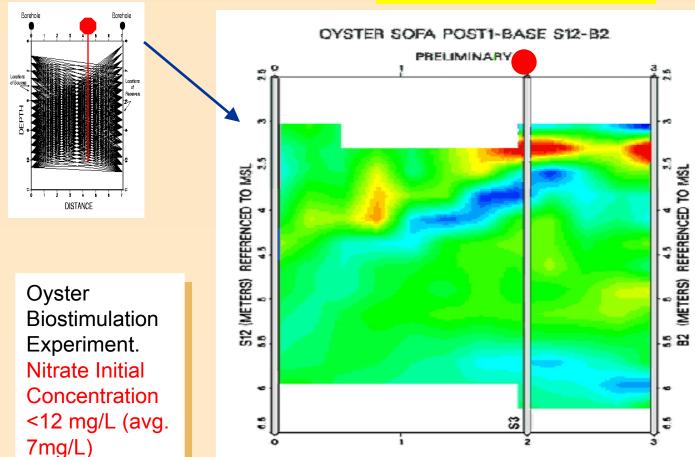
### Hydraulic Conductivity and Geophysical Responses associated with N2 gas production during Stimulation





### Preliminary Field-Scale Biostimulation Monitoring using Time-Lapse Seismic Tomography

#### **Lactate Injection Well**



13 Hours

Decrease

Variations in seismic amplitude correlated with N2 production near the wellbore AT THE FIELD SCALE



Mailloux et al.,

Change in Seismic Attenuation (blue=gas)

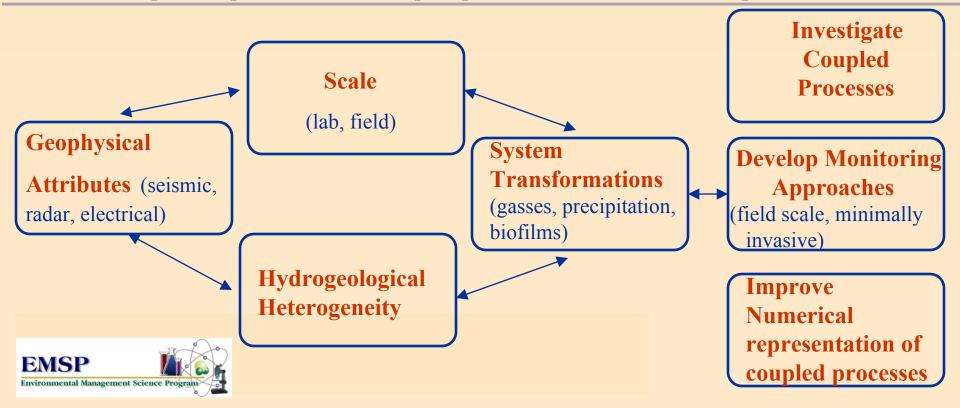
ATTENUATION (NEPERS)

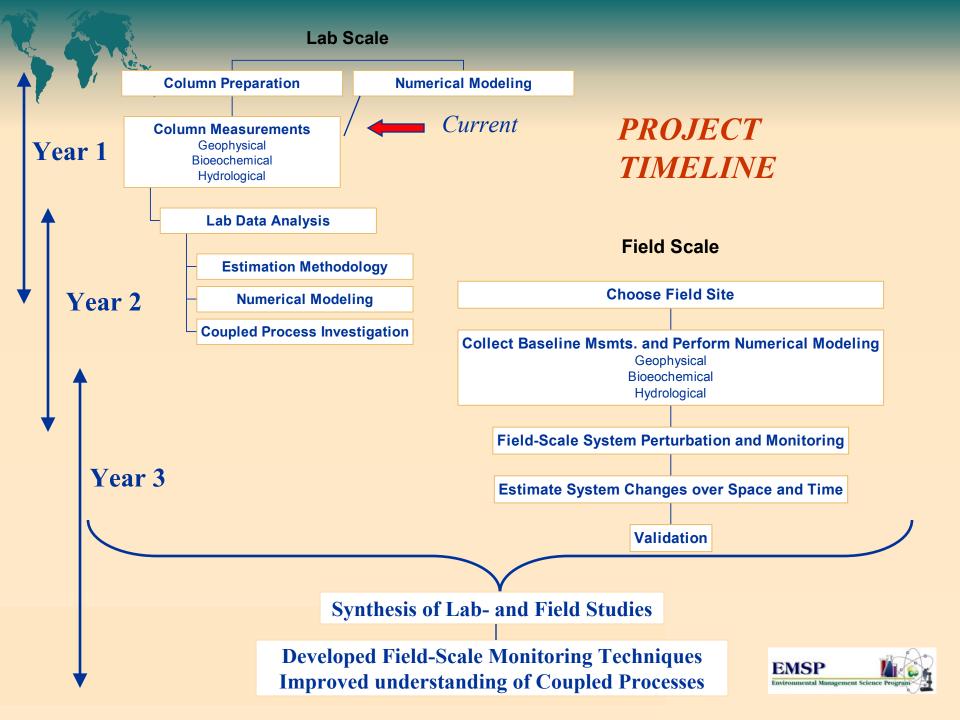
Increase

#### EMSP NEW START: RESEARCH OBJECTIVES AND APPROACH

- 1) Understand **potential of geophysical methods** to characterize and monitor hydrogeological/biogeochemical parameters/processes:
  - \* at lab and field scale
  - \* in the presence of natural heterogeneity
- 2) Use geophysical-hydrogeological-biogeochemical data to **investigate coupled processes**:
  - \*onset of correlated mineralogical-biogeochemical gradients within columns
  - \* interactions and dynamics of hydrological-biogeochemical alterations,
  - \* microbiological induced nanopartical formation and influence on flow

Additional: Improve representation of coupled processes in advanced transport codes







#### **Proposed Column Experiments**

#### **CO-LOCATED MEASUREMENTS:**

Geophysical (seismic, radar, electrical), Hydrogeological, Microbiological and Geochemical -

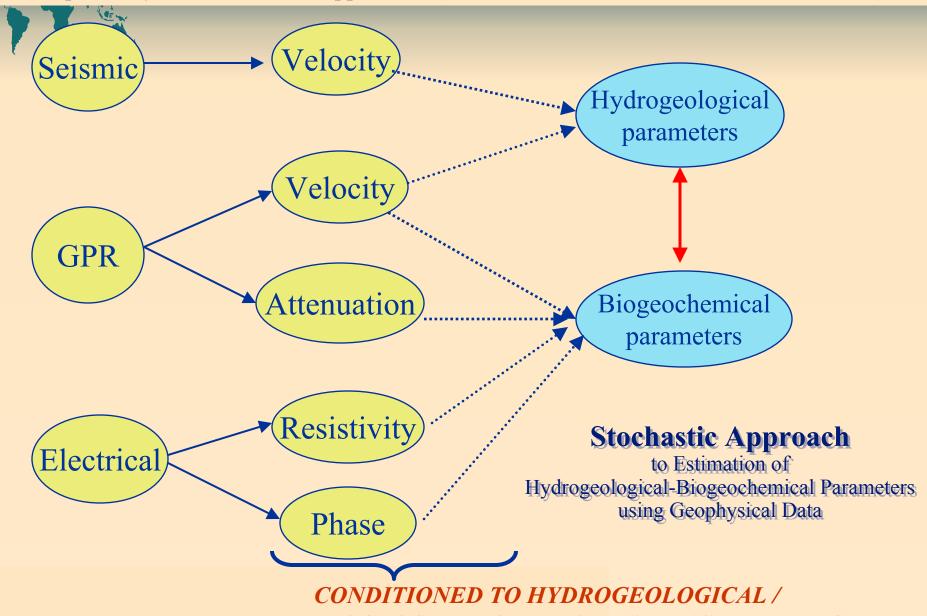
#### **EXPERIMENTS:**

| Hydrogeological<br>Heterogeneity: |       |        |         | Field Sediments/GW |               |
|-----------------------------------|-------|--------|---------|--------------------|---------------|
| Potential<br>Treatments:          | Low K | High K | Mixed K | Sieved Seds.       | Natural Seds. |
| Chemical                          | X     | X      | X       |                    |               |
| Oxidation                         |       |        |         |                    |               |
| Biostimulation to                 | X     | X      | X       | X?                 | X?            |
| induce                            |       |        |         |                    |               |
| precipitates,                     |       |        |         |                    |               |
| gasses and                        |       |        |         |                    |               |
| biofilms                          |       |        |         |                    |               |





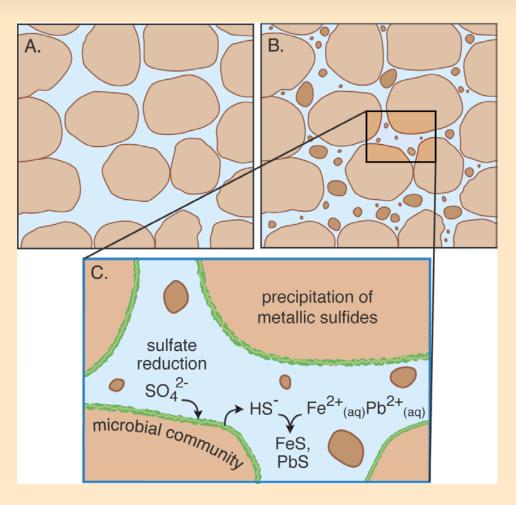
Examples: Bayesian or MCMC Approaches (Chen et al., 2002, 2001; Hubbard et al., 2001)



BIOGEOCHEMICAL "POINT" MEASUREMENTS



### **Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media**

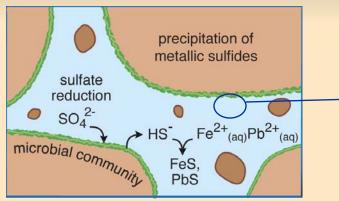


- Controlled Conditions:
- Well-defined, saturated sediments of known grain size
- Introduction of single microbial strain
- Infiltration using defined medium with fixed substrate and metal concentrations
- Fixed rate of advection





### Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media

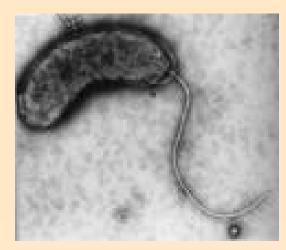




- Common soil borne and aquifer SRB
- Moderately aerotolerant; no significant reduction in cell numbers after ~4-6hr of O<sub>2</sub> exposure
- Parallel studies being conducted at LBNL/UCB (Genomes to Life)
- 0.8 x 1.0um in size; free-living growth habit in liquid culture conditions / possible biofilm formation in granular medium



60x Optical Microscope Live/Dead Image



Widdel et al., The Prokaryotes, 1992.



### **Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media**

#### **Expected Stoichiometric Reactions:**

1) SRB reduces sulfate while oxidizing lactate to acetate:

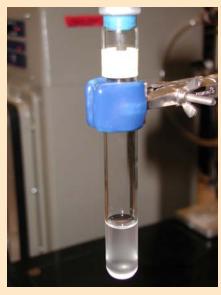
$$2CH_{3}CHOHCOO^{-} + SO_{4}^{2-} \rightarrow 2CH_{3}COO^{-} + 2HCO_{3}^{-} + 0.5H_{2}S + 0.5HS^{-} + 0.5H^{+}$$

2) Generated bisulfide reacts with metals to form metal sulfides:

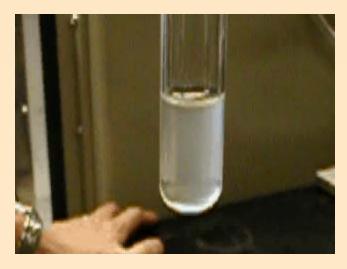
$$0.5 \text{Fe}^{2+}$$
,  $0.5 \text{Pb}^{2+} + 0.5 \text{HS}^{-} \rightarrow 0.5 \text{FeS}$ , PbS +  $0.5 \text{H}^{+}$ 

Composite Reaction:

$$2\text{CH}_{3}\text{CHOHCOO}^{-} + \text{SO}_{4}^{2-} + 0.5\text{Fe}^{2+}, 0.5\text{Pb}^{2+} \rightarrow 2\text{CH}_{3}\text{COO}^{-} + 0.5\text{FeS}, \text{PbS} + 2\text{HCO}_{3}^{-} + 0.5\text{H}_{2}\text{S} + \text{H}^{+}$$



Liquid culture of *D. vulgaris* after 3 days on defined medium (~10<sup>8</sup> cells/mL)



Addition of 70ppm Fe<sup>2+</sup> (aq)

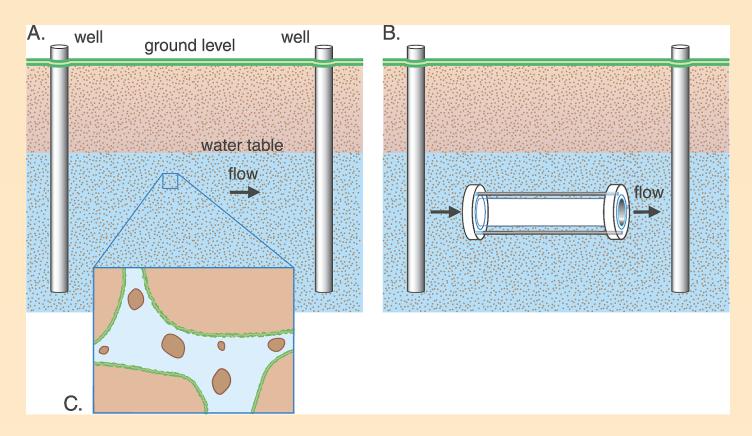
Rapid production of Metal Sulfides under Stimulated Conditions!!

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### **Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media**

#### **Mimic Subsurface Using Flow-Through Column Experiments:**

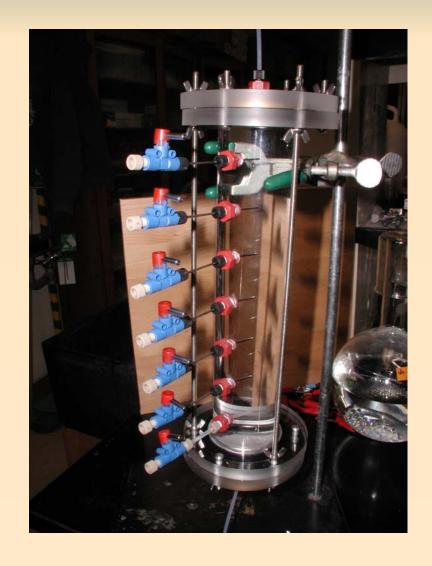






### Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media

- Simultaneous Measurements on Control and Innoculated Columns: Biogeochemical, Hydrological and Geophysical
  - **\*** Multiport Biogeochemical Sampling:
    - Fluid chemistry (anions, cations, metals, pH) using ion chromatography, ion coupled plasma, etc.
    - Biomass sampling (quantitative live/dead, PLFA)
    - Nanoparticle and/or flocculate sampling and analysis (XRD, SEM, TEM)
  - Chemical results directly feed modeling effort





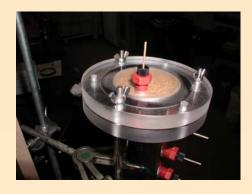


### Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media

- Simultaneous Column Experiments (cont'd.)
  - \* Hydraulic Conductivity
  - **\*** Geophysical Measurements:
    - Radar
    - Seismic
    - Electrical
      - Complex Resistivity (1-10<sup>5</sup>Hz) and Streaming potentials
        - Non-polarizing (i.e. lownoise) AgCl potential electrodes
        - Gold current electrodes











### **Experiment Series #1: Microbe-Induced Sulfide Precipitation in Porous Granular Media**

#### Preliminary Column Test:

- Flow-through of innoculum
- Advection rate = 7cm/day
- Non-diluted growth medium with 10ppm Fe<sup>2+</sup>

#### Observations

- Noticeable black precipitate (FeS) after only 2-3 days
- Gradational precipitation front
- Slight reduction in effluent pH (7.1 to 6.8)
- Gas generation (CO<sub>2</sub>, acetate?)

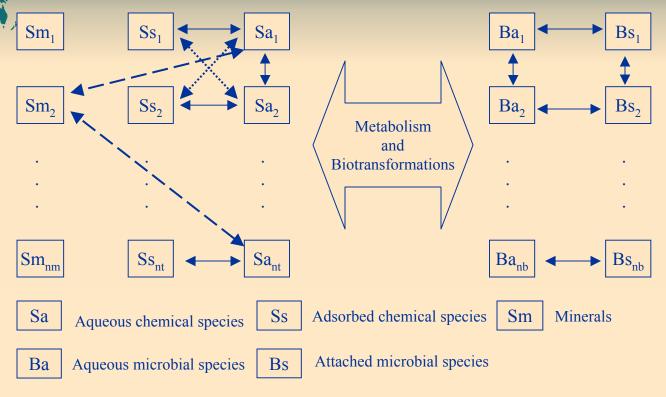




#### **♦** Forthcoming Column Tests:

- Seismic Measurements
  - Decreases expected in amplitudes due to gases; increases expected in velocity due to increasing bulk density (mineral precipitates)
- · Radar Measurements
  - Decreases in dielectric associated with gases and possibly sulfides
- Electrical methods
  - ▶ Changing complex electrical response and streaming potentials; increasing apparent resistivity due to precipitation of *polarizable* minerals (sulfides) and more resistive oxides
- Hydraulic Measurement
  - Pore throat plugging (gases, minerals)
- Biogeochemical Measurements
  - refer to modeling results...

#### **Mechanics of Coupled Numerical Model BIOCORE (Zhang, 2000)**



- Mass transfer between phases of given species: adsorption/desorption of chemical species and attachment/detachment of microbial species
- Reactions among aqueous chemical species: acid-base, redox and complexations reactions and competition, metabiosis and endogeneous respiration among microbial species
- Mass transfer between phases of different species: ion exchange
  - Mass transfer between phases: mineral dissolution-precipitation



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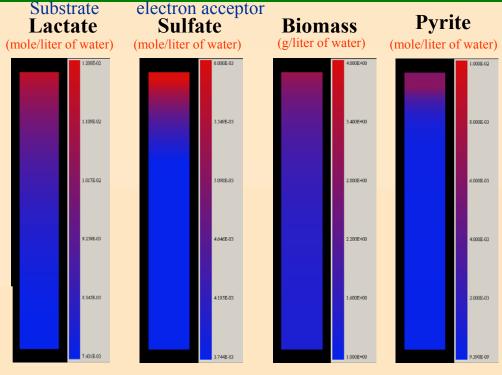


# Pre-Experimental Biogeochemical Interaction Simulations using BIOCORE

At 4 days: (steady state reached at ~30 days)

#### **Species** Microbes Desulfovibrio vulgaris 16 Primary H<sub>2</sub>O, H<sup>+</sup>, Citrate<sup>-</sup>, Lactate<sup>-</sup>, SO4<sup>-2</sup>, HCO<sub>3</sub><sup>-</sup>, SiO<sub>2</sub>(aq), HPO<sub>4</sub><sup>-2</sup>, CΓ, NO<sub>3</sub><sup>-</sup>, Aqueous $Na^+, K^+, Ca^{+2}, Mg^{+2}, Fe^{+2}, O_2(aq)$ **Species** HS<sup>-</sup>, H<sub>2</sub>S(aq), MgHPO<sub>4</sub>(aq), CO<sub>2</sub>(aq), MgHCO<sub>3</sub><sup>+</sup>, NaHCO<sub>3</sub>(aq), CaHPO<sub>4</sub>(aq), H<sub>2</sub>PO<sub>4</sub>, CaHCO<sub>3</sub>, FeHCO<sub>3</sub>, CaNO<sub>3</sub>, NaHPO<sub>4</sub>, MgPO<sub>4</sub>, Methane(aq), NaCl(aq), MgCl<sup>+</sup>, FeHPO<sub>4</sub>(aq), MgCO<sub>3</sub>(aq), CaCO<sub>3</sub>(aq), CaPO<sub>4</sub><sup>-</sup>, CO<sub>3</sub><sup>-2</sup>, FePO<sub>4</sub><sup>-</sup>, FeCO3(aq), NaHSiO<sub>3</sub>(aq), HSiO<sub>3</sub><sup>-</sup>, 62 CaCl<sup>+</sup>, Acetic acid(aq), MgP2O<sub>7</sub><sup>-2</sup>, NaCO<sub>3</sub><sup>-</sup>, MgSO4(aq), OH<sup>-</sup>, FeOH<sup>+</sup>, Secondary NaSO<sub>4</sub><sup>-</sup>, FeCl<sup>+</sup>, CaSO<sub>4</sub>(aq), CaP<sub>2</sub>O<sub>7</sub><sup>-2</sup>, S<sup>-2</sup>, PO<sub>4</sub><sup>-3</sup>, CaCL<sub>2</sub>(aq), CaOH<sup>+</sup>, S<sub>2</sub><sup>-2</sup>, FeSO<sub>4</sub>(aq), H<sub>2</sub>(aq), NaOH(aq), HP<sub>2</sub>O<sub>7</sub><sup>-3</sup>, S<sub>3</sub><sup>-2</sup>, Acetate<sup>-</sup>, NaHP<sub>2</sub>O<sub>7</sub><sup>-2</sup>, H<sub>3</sub>PO<sub>4</sub>(aq), NaP<sub>2</sub>O<sub>7</sub><sup>-3</sup>, Na<sub>2</sub>P<sub>2</sub>O<sub>7</sub><sup>-2</sup>, P<sub>2</sub>O<sub>7</sub><sup>-4</sup>, Fe(OH)<sub>2</sub>(aq), H<sub>2</sub>P<sub>2</sub>O<sub>7</sub><sup>-2</sup>, S<sub>4</sub><sup>-2</sup>, Aqueous Species HCl(aq), MgCH<sub>3</sub>COO<sup>+</sup>, H<sub>2</sub>SiO<sub>4</sub><sup>-2</sup>, H<sub>6</sub>(H<sub>2</sub>SiO<sub>4</sub>)<sub>4</sub><sup>-2</sup>, S<sub>2</sub>O<sub>3</sub><sup>-2</sup>, NaCH<sub>3</sub>COO(aq). $Fe^{+3}$

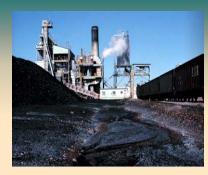
7 Minerals Siderite, Calcite, Pyrite, Quartz, Dolomite, Greenalite, Troilite,





#### **Forthcoming Field Studies**

D-Area Coal Pile Runoff Basin, SRS.
 Characteristics:



- ❖ Contamination: Runoff from coal storage produced contaminants (sulfate, Al, Fe, Pb,Mg, Zn..~1/2 mile from the Savannah River). Percolation of acid-sulfate waters are key concern.
- Geology: Heterogeneous interbedded sand, silt, clay layers (porous granular).
- Hydrology: Water table 0-15ft BGS, aquifer of interest ~45-55' BGS.
- Remediation: Many remediation approaches are being considered, including sulfate reduction to immobilize metals via stimulation (Charles Turick, Michael Heitkamp, SRS).

#### Other Sites?







#### **Project Summary and Expected Results**

#### **PROJECT SUMMARY**

- ◆ Components: Theoretical, numerical, stochastic, and experimental
- ◆ Interdisciplinary: Hydrogeology-Geophysics-Biogeochemistry-Statistics
- ◆ Multiple Scales of Investigation: Lab and Field

#### **EXPECTED RESULTS**

- ◆ Potential of using **geophysical methods as a minimally invasive, cost- effective, and field-scale approach** for investigating and monitoring system transformations;
- Investigate Coupled processes that influence contaminant transport and remediation;
- ◆ Improved representation of coupled biogeochemical-hydrological processes in advanced numerical transport models;
- ◆ Assist in assessing the efficacy of remediation at a DOE field site.







#### The end



